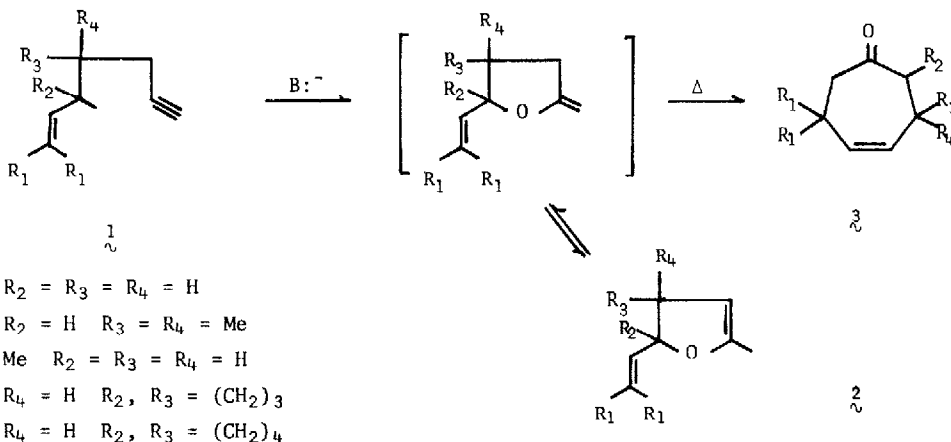


A NOVEL SYNTHESIS OF 4-CYCLOHEPTENONES

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4-Cycloheptenones, mono and bicyclic, can be prepared by a base catalyzed thermal reaction of 1-hepten-6-yn-3-ols.

A broad variety of sesquiterpenes, mono and bicyclic, either incorporate or are readily accessible from compounds having a 4-cycloheptenone moiety.¹ It appeared that the synthesis of Demole and Enggist,² studied in detail by Rhoads,³ would be very useful if generalized. The major problem rests in the preparation of 2-methylene-5-vinyltetrahydrofurans. We describe here a novel solution to the problem, its advantages and some limitations. The approach is dependent on the base catalyzed cyclization of 4-pentyn-1-ols first described by Paul and Tchelitcheff⁴ and exploited further by Eglinton, Jones and Whiting.⁵ This cyclization uses



catalytic amounts (3-5%) of a strong base (NaH) and since an added vinyl group should not interfere the resulting 2-methylene-5-vinyltetrahydrofuran should be thermally convertible to a 4-cycloheptenone. In some cases the process can be accomplished in a single step while in others it may be preferable to carry out the reaction in two steps.

For example $1a$ shows no tendency to form $2a$ when heated at 125° for 8 hrs., but is converted in 30% yield (not optimized) to $3a$ at 170° for 24 hrs. Conversely $1b$ is readily converted to $2b$ in 80% yield under the milder conditions above, and $2b$ gives $3b$ (81%) when heated at 225° in quinoline for 24 hrs. Presence of two methyls on the terminal carbon of the vinyl group prevents the reaction entirely. Bicyclic compounds $1d$ and $1e$ are converted to $3d$ and $3e$ in a single step in diphenyl ether solution. At 195° for 12 hrs. $1d$ gives $3d$ in 51% yield, and at 185° (12 hrs) $1e$ is converted to $3e$ in 78% yield. We attribute the failure of $1c$ to react to steric inhibition of the Claisen rearrangement step, but otherwise the scheme is adapted to the preparation of a wide variety of substituted cycloheptenones.

The single step procedure is simple and convenient. For example a solution containing $1g$ (6.1 mmol) of $1g$ in 3 ml of diphenyl ether was placed in a 10 ml round-bottomed flask under nitrogen. After 0.19 ml of 1.6 M methylolithium in ether (0.3 mmol) had been added, the mixture was stirred at 185° for 12 hrs. The product was separated on silica gel (pentane/ether) and the crude liquid was distilled in a Kugelrohr giving 0.84 g (84%) of $3g$, NMR (CCl_4) δ 1.30 (m, 3H) 1.78(m, 8H), 2.26(m, 5H), 2.68(m, 2H) 5.54(m, 1H).

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